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Chemistry B11 Bakersfield college Chemistry B11 Chapter 10

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Acids and bases Arrhenius definitions: an acid is a substance that produces H_3O^+ ions in aqueous solution. A base is a substance that produces OH^- ions in aqueous solution. $CH_3COOH(aq) + H_2O(l) \rightleftharpoons CH_3COO^-(aq) + H_3O^+(aq)$ CH_3COOH is an acid. $NH_3(aq) + H_2O(l) \rightleftharpoons NH_4^+(aq) + OH^-(aq)$ NH_3 is a base.

Chemistry B11 Chapter 10 Acids and bases

Chemistry B11 Bakersfield College Chemistry B11 Chapters 13 Esters, amides and carbohydrates Esters: esters are derived from carboxylic acids (the hydrogen atom in the carboxyl group of carboxylic acid is replaced by an alkyl group). The functional group of an ester is a carbonyl group bonded to an $-OR$ group.

Chemistry B11 Chapters 13 Esters, amides and carbohydrates

Chemistry B11 Bakersfield college Chemistry B11 Chapters 10-11 Alkanes, Alkenes, Alkynes and Benzene Organic compounds: organic chemistry is the chemistry of carbon and only a few other elements-chieflly, hydrogen, oxygen, nitrogen, sulfur, halogens, and phosphorus (from 116 elements).

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derivatives from ammonia (NH₃). Aliphatic amines: an amine in which nitrogen is bonded only to alkyl group or hydrogens. Aromatic amines: an amine in which nitrogen is bonded to one or more aromatic rings. Note: amines are classified as primary (°°°1), secondary ...

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solutions have some properties: 1. The distribution of particles in a solution is uniform. Every part of the solution has exactly the same composition and properties as every other part (a single phase-homogenous). 2.

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