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Write a balanced chemical equation and use the necessary symbols from Table 11.1 to describe the reaction completely: Bubbling chlorine gas through a solution of potassium iodide gives elemental iodine and a solution of potassium chloride

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Chapter 11 Assessment Chemistry Answer Key

Solutions Manual Chemistry: Matter and Change • Chapter 11 211 CHAPTER 11 SOLUTIONS MANUAL Section 11.1 Assessment page 372 5. Compare the mass of the reactants and the mass of the products in a chemical reaction, and explain how these masses are related. The coefficients in the balanced equation indicate the molar relationship between each pair of

StoichiometryStoichiometry

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Mastering Chemistry - Chapter 11 Convert Between Units of Concentration Molarity is a common unit of concentration because the volume of a liquid is very easy to measure. However, the drawback of using molarity is that volume is a temperature-dependent quantity. As temperature changes, density changes, which affects volume.

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370 Chapter 11 • Stoichiometry EXAMPLE Problem 11.1 Interpreting Chemical Equations The combustion of propane (C_3H_8) provides energy for heating homes, cooking food, and soldering metal parts. Interpret the equation for the combustion of propane in terms of representative particles, moles, and mass.

Chapter 11: Stoichiometry

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