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General Chemistry 10th Edition by Darrell Ebbing (Author), Steven D. Gammon (Author) 4.1 out of 5 stars 65 ratings. ISBN-13: 978-1285051376. Darrell Ebbing has taught general chemistry for more than thirty years and is now retired from Wayne State University. He received his Ph.D. in physical chemistry from Indiana University.

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Darrell Ebbing has taught general chemistry for more than thirty years and is now retired from Wayne State University. He received his Ph.D. in physical chemistry from Indiana University. Steven D. Gammon is the Provost at the University of Maine Fort Kent.

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Darrell Ebbing's GENERAL CHEMISTRY is recognized for its signature clear explanations, macro to micro orientation, and enhanced problem-solving strategies that have made it a best seller. The eleventh edition incorporates nomenclature changes and the adoption of IUPAC periodic table conventions, along with changes to the discussion of d orbitals.

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MSC: general chemistry 16. Which of the following compounds would be expected to have the highest melting point? A) NCl 3 B) OCl 2 C) MgCl 2 D) LiCl E) CCl 4 ANS: C PTS: 1 DIF: moderate REF: 9.1 OBJ: Describe some general properties of ionic substances. TOP: bonding | ionic bonding KEY: properties of ionic substance MSC: general chemistry 17.

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The eleventh edition was carefully reviewed with an eye toward strengthening the content available in OWLv2, end-of-chapter questions, and updating the presentation. Nomenclature changes and the adoption of IUPAC periodic table conventions are highlights of the narrative revisions, along with changes to the discussion of d orbitals. In-text examples have been reformatted to facilitate learning ...

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